

Identifying Pd-Based Ternary Membranes for Carbon and Sulfur Applications

Amanda Lewis, Hongbin Zhao,
& Scott Hopkins

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Outline

- ❑ Project overview
- ❑ Palladium membrane technology background
- ❑ Current project progress
- ❑ Future plans



Project Overview

Funding

- \$1,517,000 Total
- \$1,207,000 U.S. Department of Energy
- \$310,000 Cost Share

Performance Period

- October 1, 2009 to September 30, 2012 (three-year project as proposed)
- No cost extension(s) to September 30, 2014

Participants

- Pall Corporation (project management, support fabrication, module construction, mixed-gas testing)
- Cornell University (composition spread fabrication on silicon wafer)
- Georgia Institute of Technology (surface characterization of exposed wafers)
- Colorado School of Mines (alloy membrane fabrication and annealing)



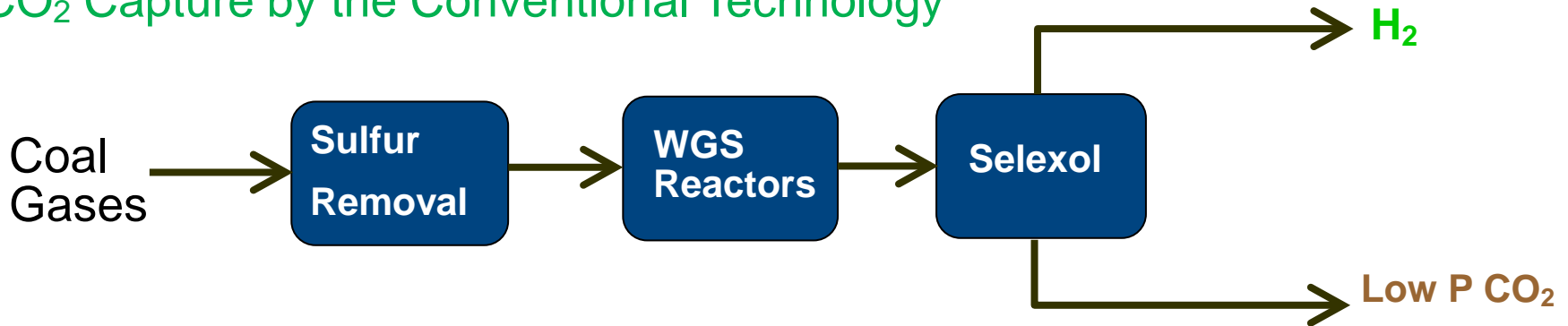
Project Overview Continued

Project Objectives

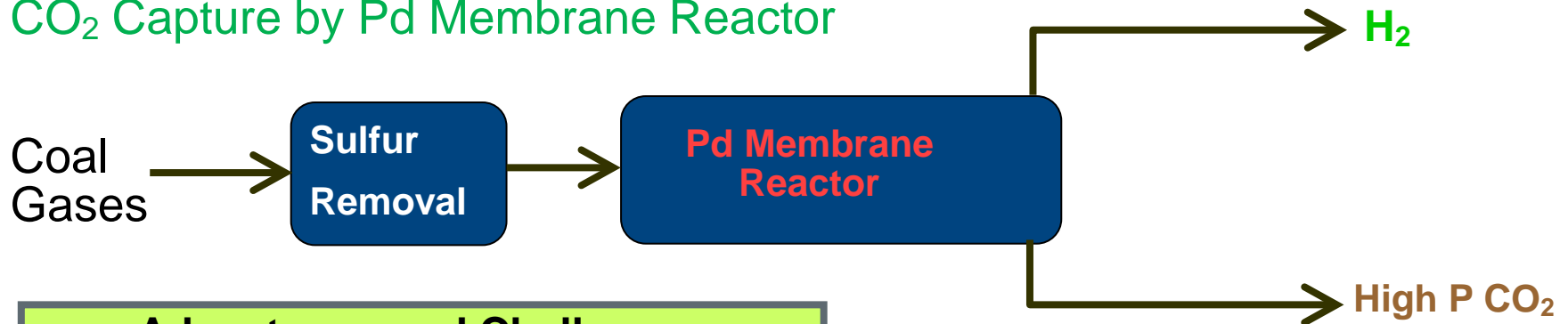
- ❑ Develop a high temperature and pressure membrane based system for CO₂ capture and hydrogen production that resists moderate levels of contaminants, typical in gasified coal.
- ❑ Identify chemically resistant palladium-based ternary alloy(s) using a high-throughput screening method.
- ❑ Demonstrate lab-scale fabrication and testing reproducibility.
- ❑ Understand long-term effects of carbon monoxide and hydrogen sulfide before scaling up.

Current Technology

CO₂ Capture by the Conventional Technology



CO₂ Capture by Pd Membrane Reactor



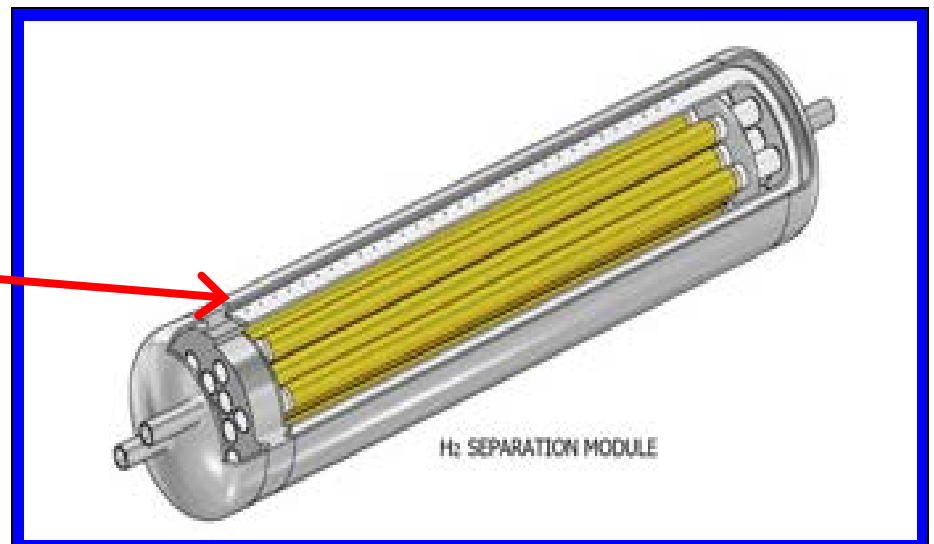
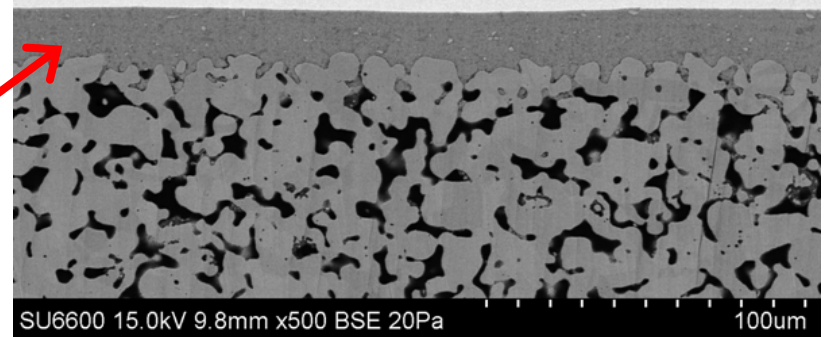
Advantages and Challenges

- Offer the potential for combined processes
- Reduce CO₂ compression costs
- Severe poisoning of membrane surface by sulfur and others

Pall's Palladium Membrane Technology



Palladium Membrane (3-5 microns thick)



Why Palladium

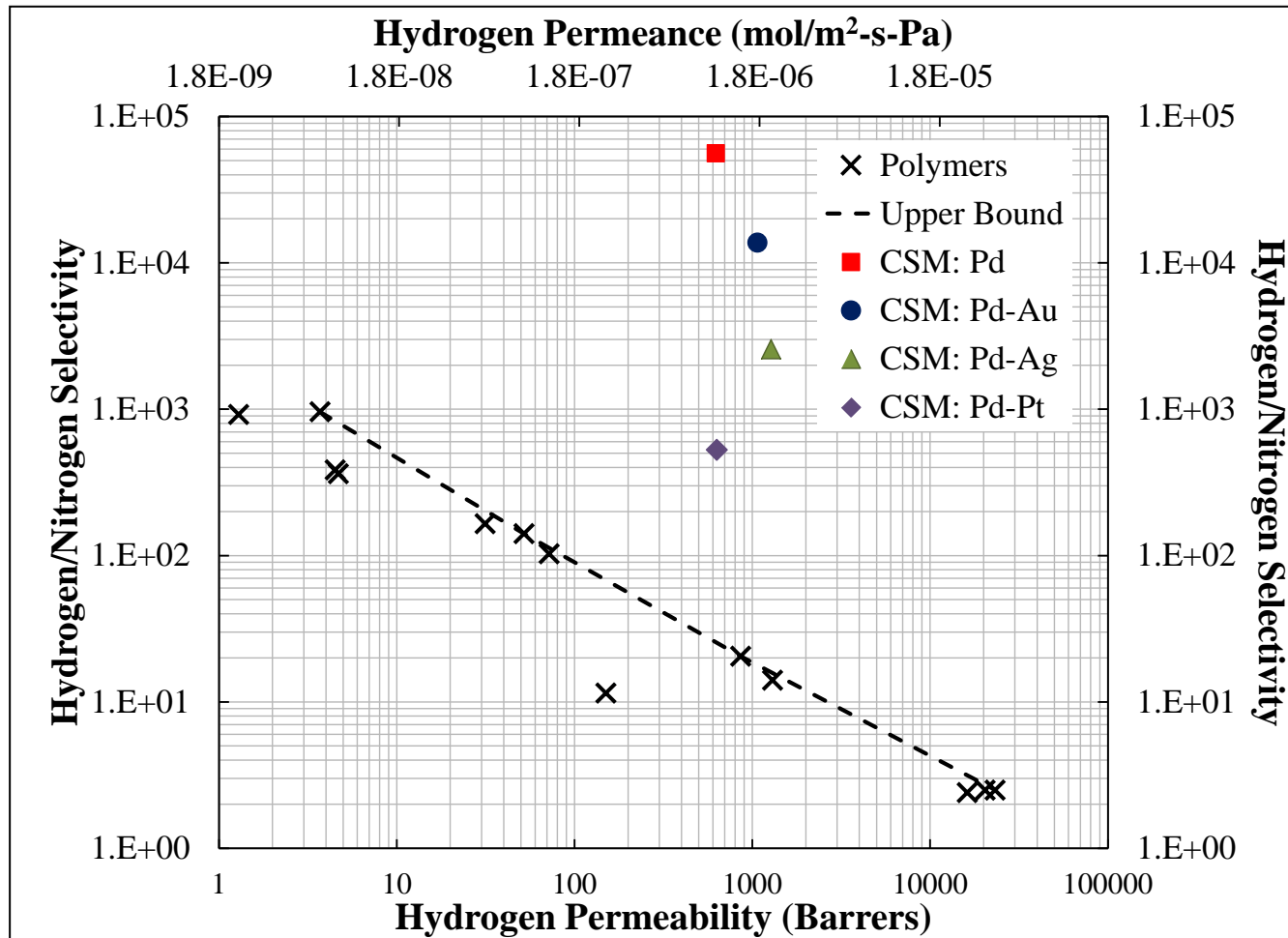


Figure adapted from:

1. Robeson, L.M. *Journal of Membrane Science*. **62**(2):(1991) p. 165-185.
2. Robeson, L.M. *Journal of Membrane Science*. **320**(1-2):(2008) p. 390-400.
3. Lewis, A. (2012) "How Contaminants and Alloying Metals Affect Hydrogen Permeation in Pd-Based Composite Membranes" (Ph.D. Thesis) – Colorado School of Mines.

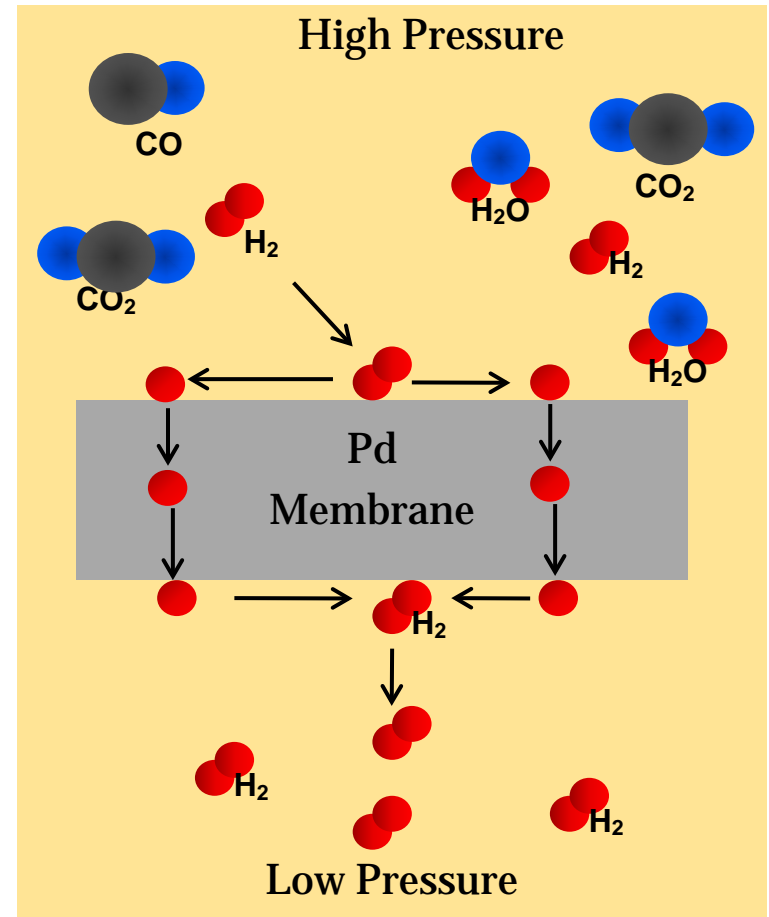
How Palladium Membrane Works

Permeation

1. Dissociative Adsorption
2. Surface-to-bulk
3. Bulk Diffusion
4. Bulk-to-Surface
5. Recombinative Desorption

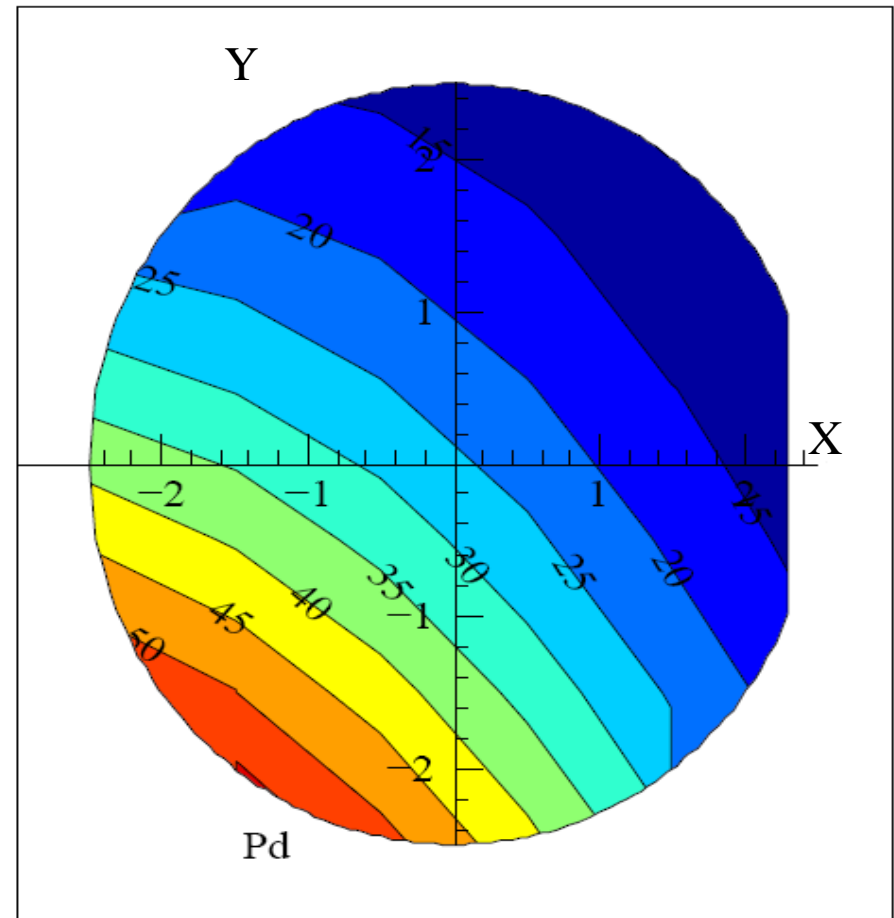
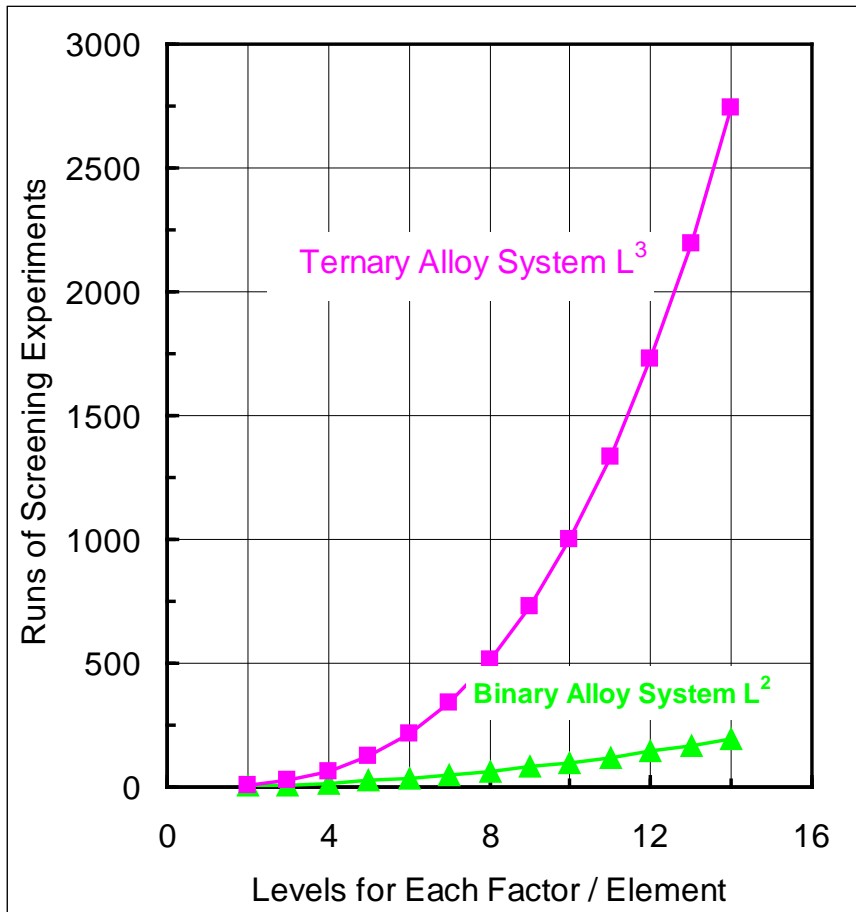
Disadvantages

1. Poor chemical resistance
2. Embrittlement



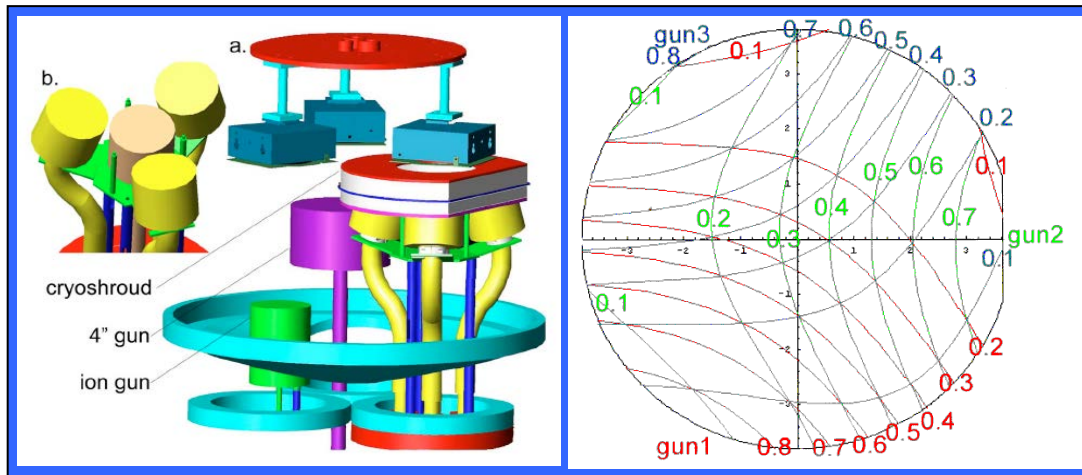
High Throughput Alloy Screening

Conventional: One at a Time vs. Combinatorial: Hundreds at a Time

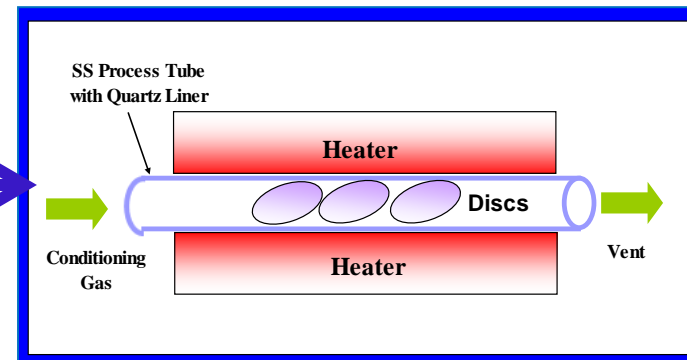


Combinatorial Pd Alloy Development Workflow

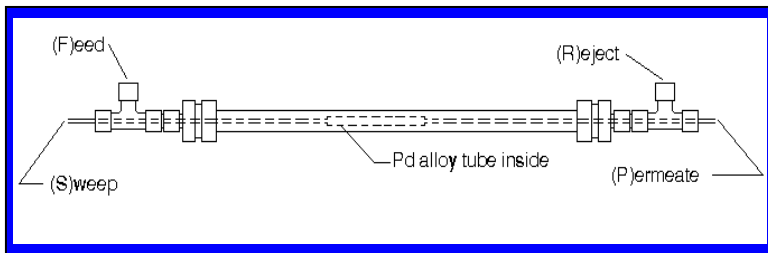
Composition Spread Fabrication: 50 alloys Prof. Bruce van Dover (Cornell)



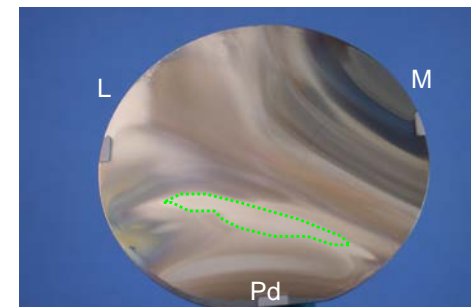
Syngas Exposure: Pall Corporation



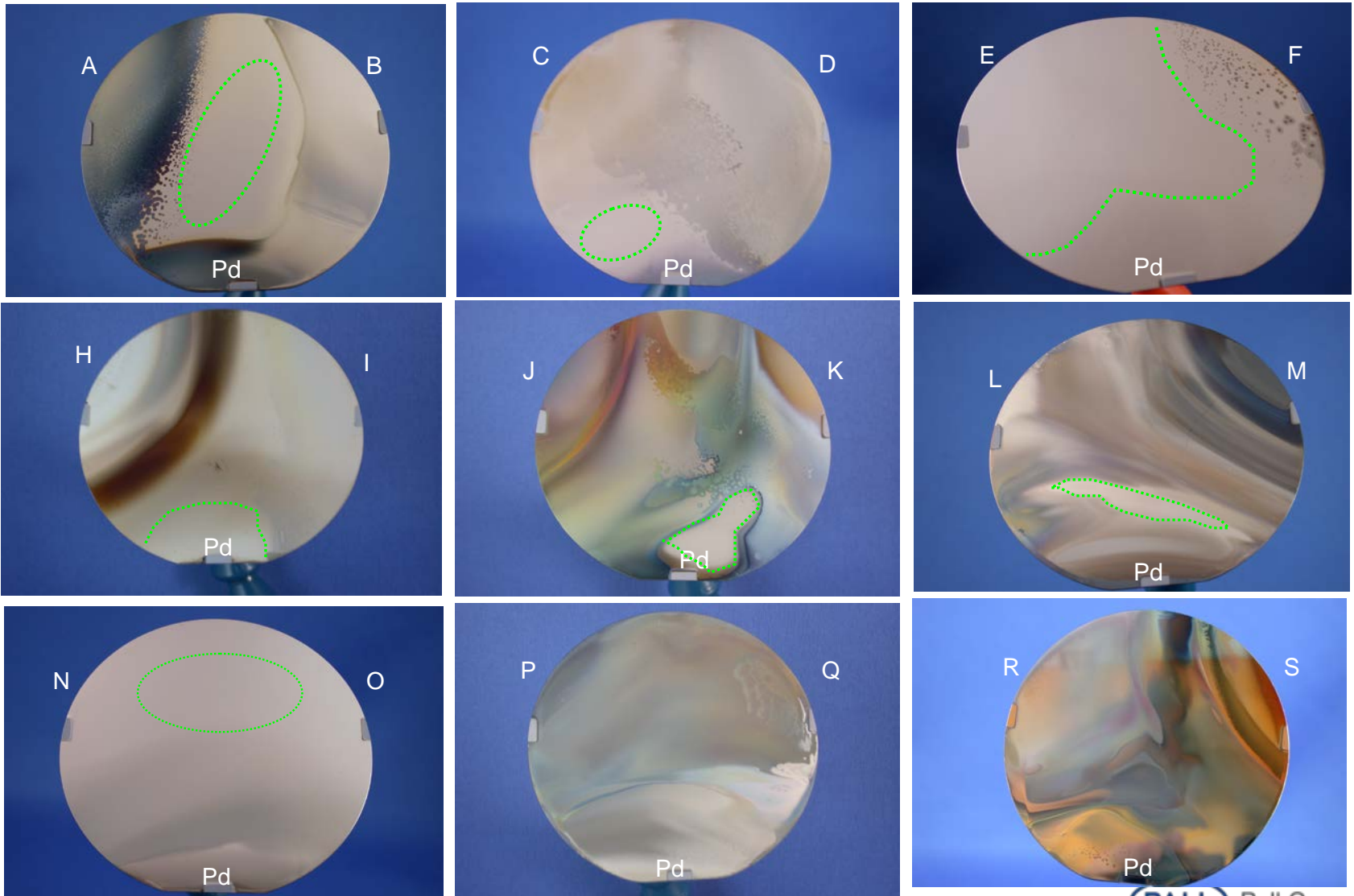
Fabrication and Testing: 6 alloys Prof. Doug Way (CSM) and Pall



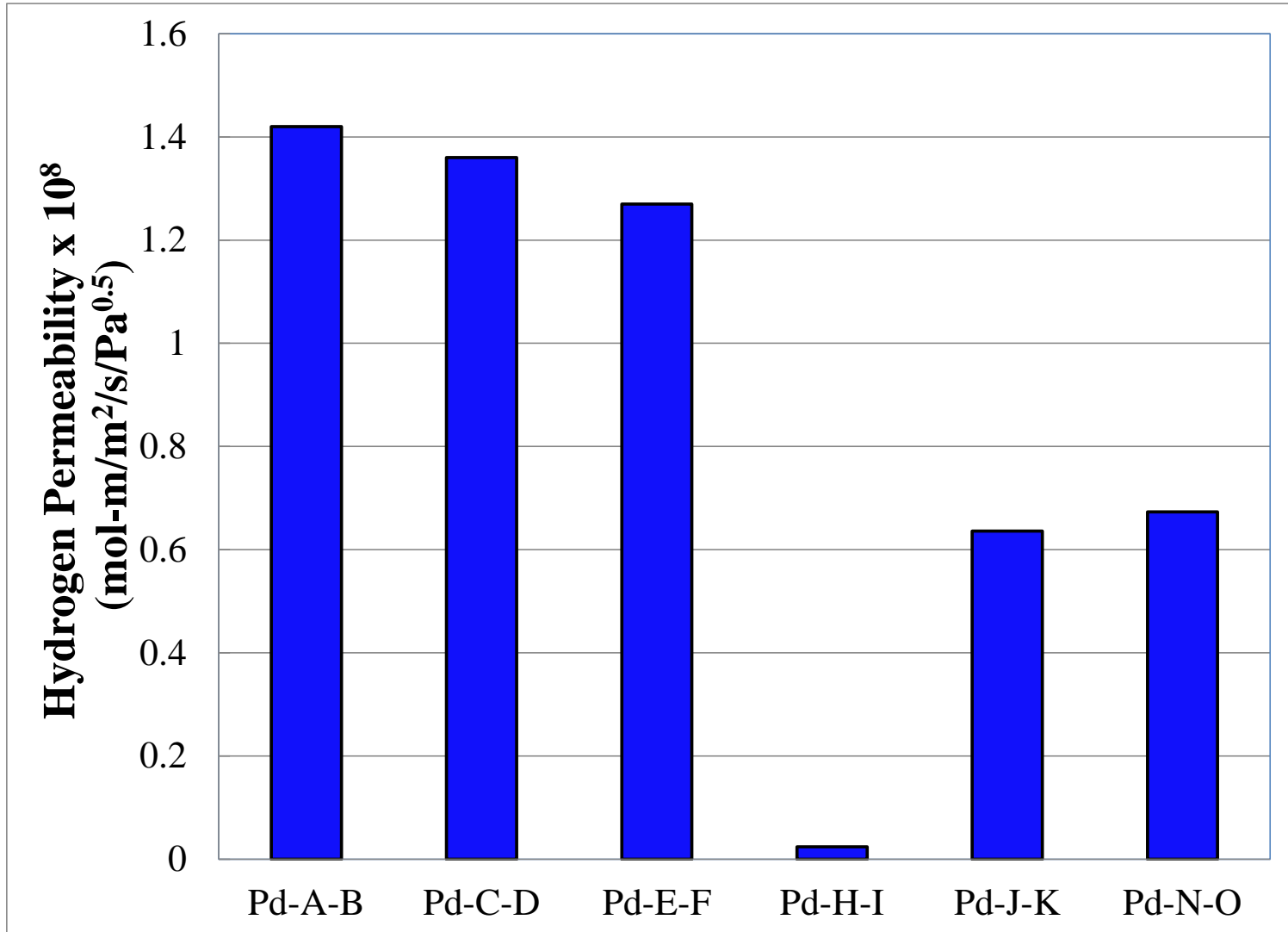
Corrosion Assessment: Prof. Meilin Liu (Georgia Tech)



Sputtered Wafers After Syngas Exposure



Pure-gas Testing Summary at 500°C



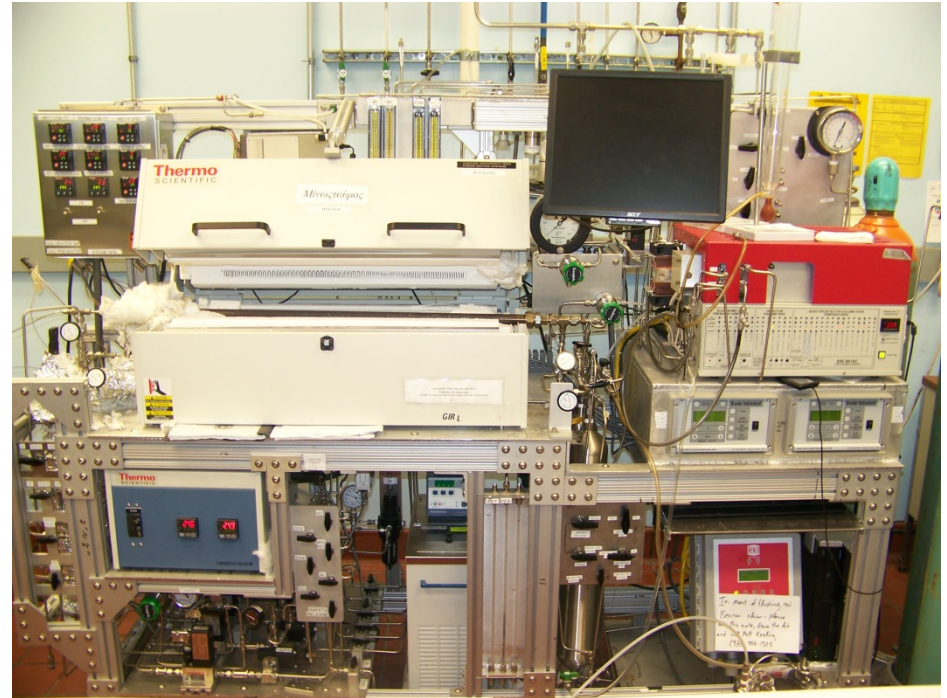
Pd Alloy Simulated Syngas Testing

Testing Conditions

- ❑ Feed Pressure ~160 psig
- ❑ Temperature 400 / 500°C

Testing Sequence

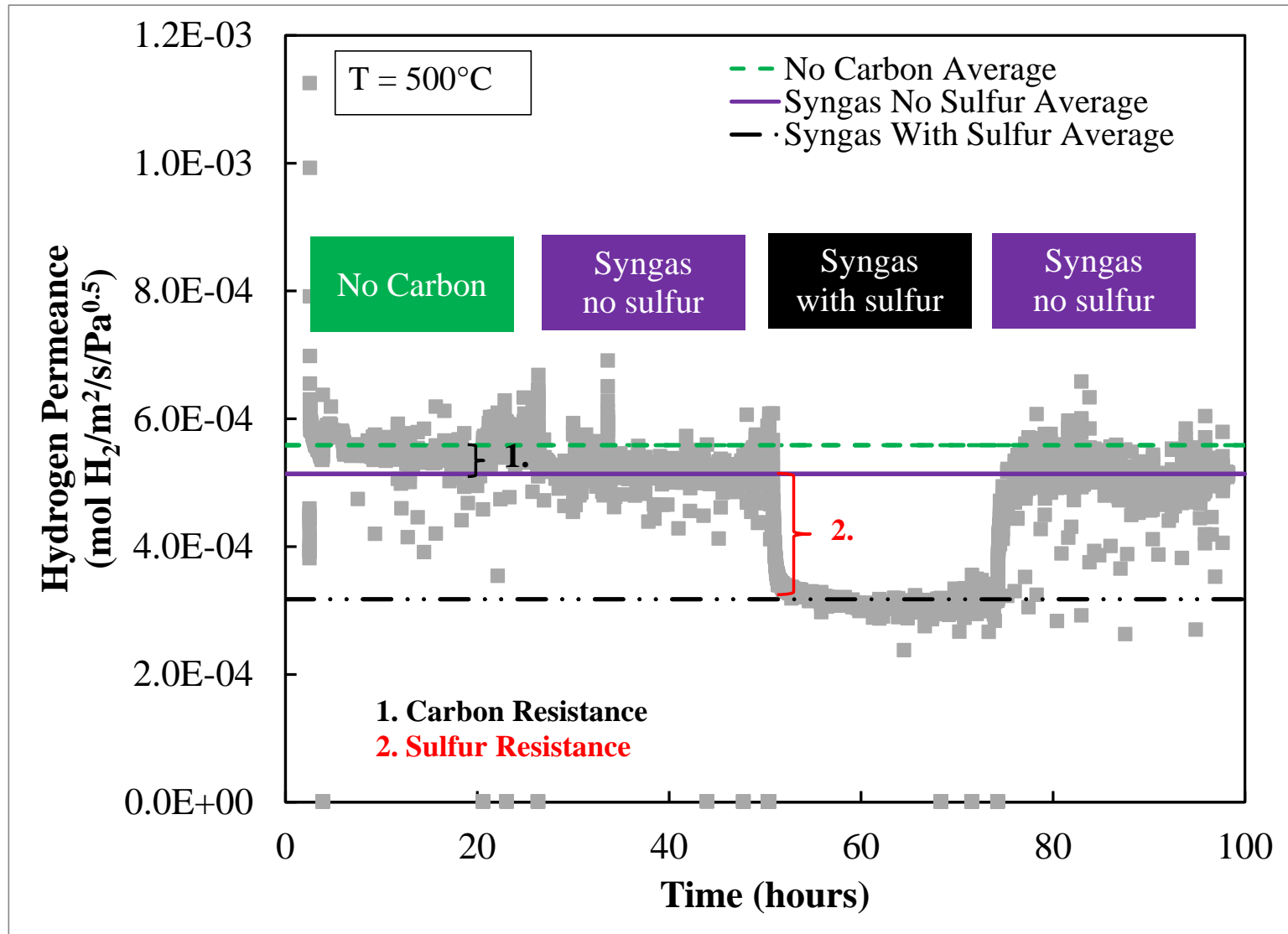
- ❑ 24 hours in 36 v% H₂, 3 v% H₂O, balance N₂
- ❑ 24 hours in syngas
- ❑ 24 hours in syngas + 20 ppm H₂S
- ❑ 24 hours in syngas



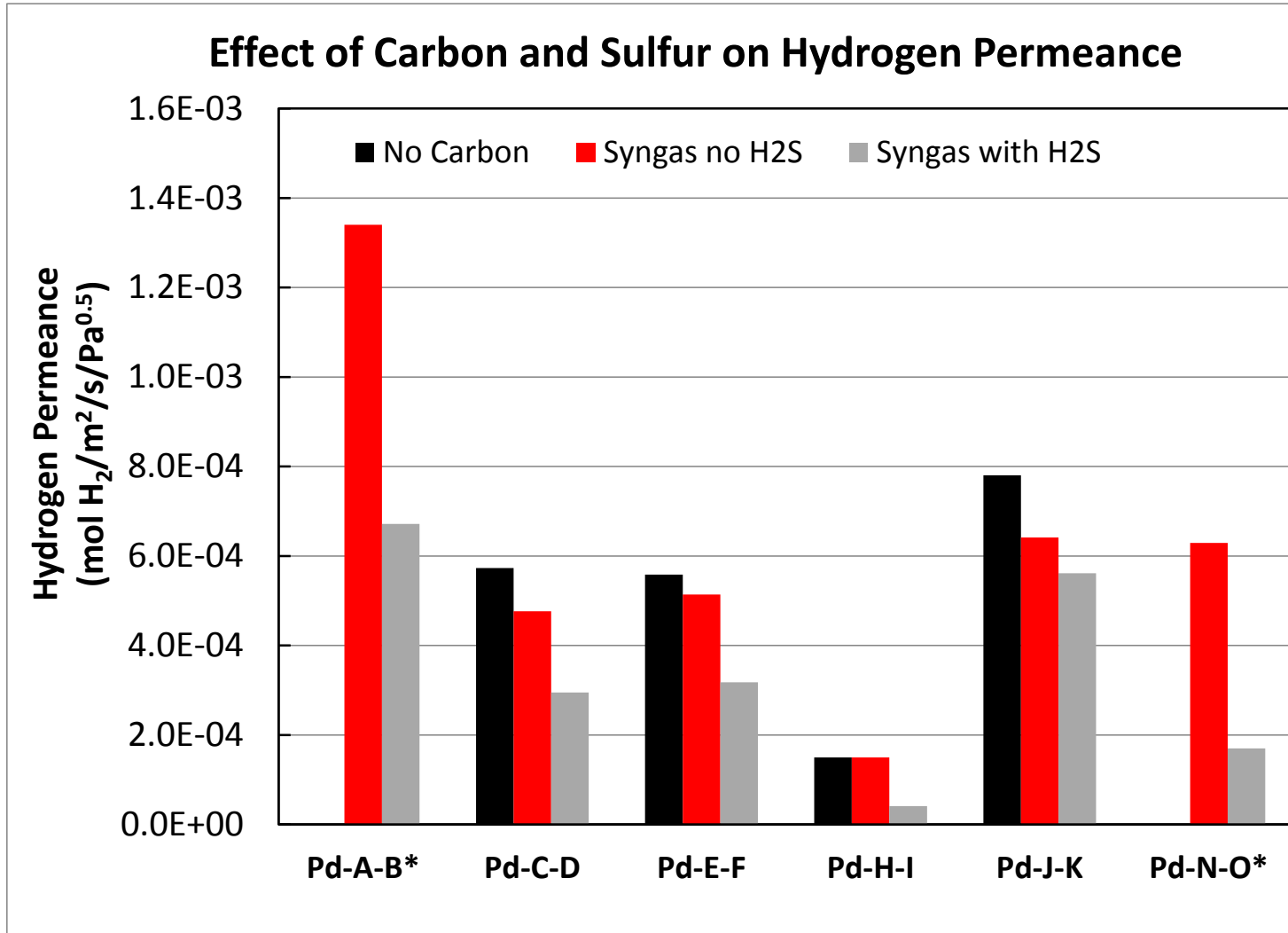
Syngas (air blown coal gasifier)

- ❑ H₂ = 36 v%
- ❑ CO₂ = 11 v%
- ❑ CO = 1.3 v%
- ❑ H₂O = 3 v%
- ❑ N₂ = 49 v%

Sample Test Sequence

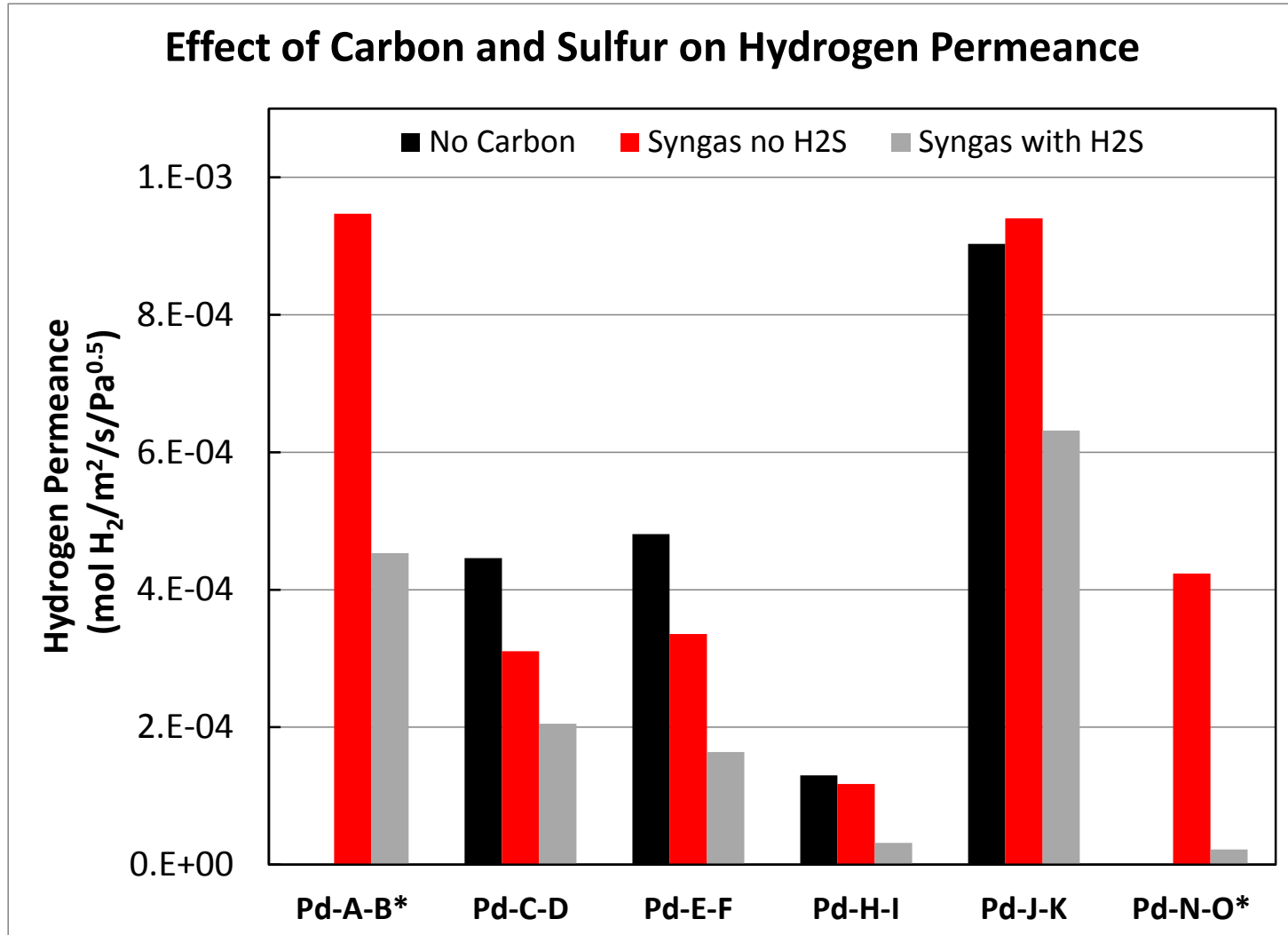


Testing Results at 500°C, $P_{\text{feed}} = 160$ psig



*Note: membranes without the “no carbon” data were tested before the current testing plan was in place.

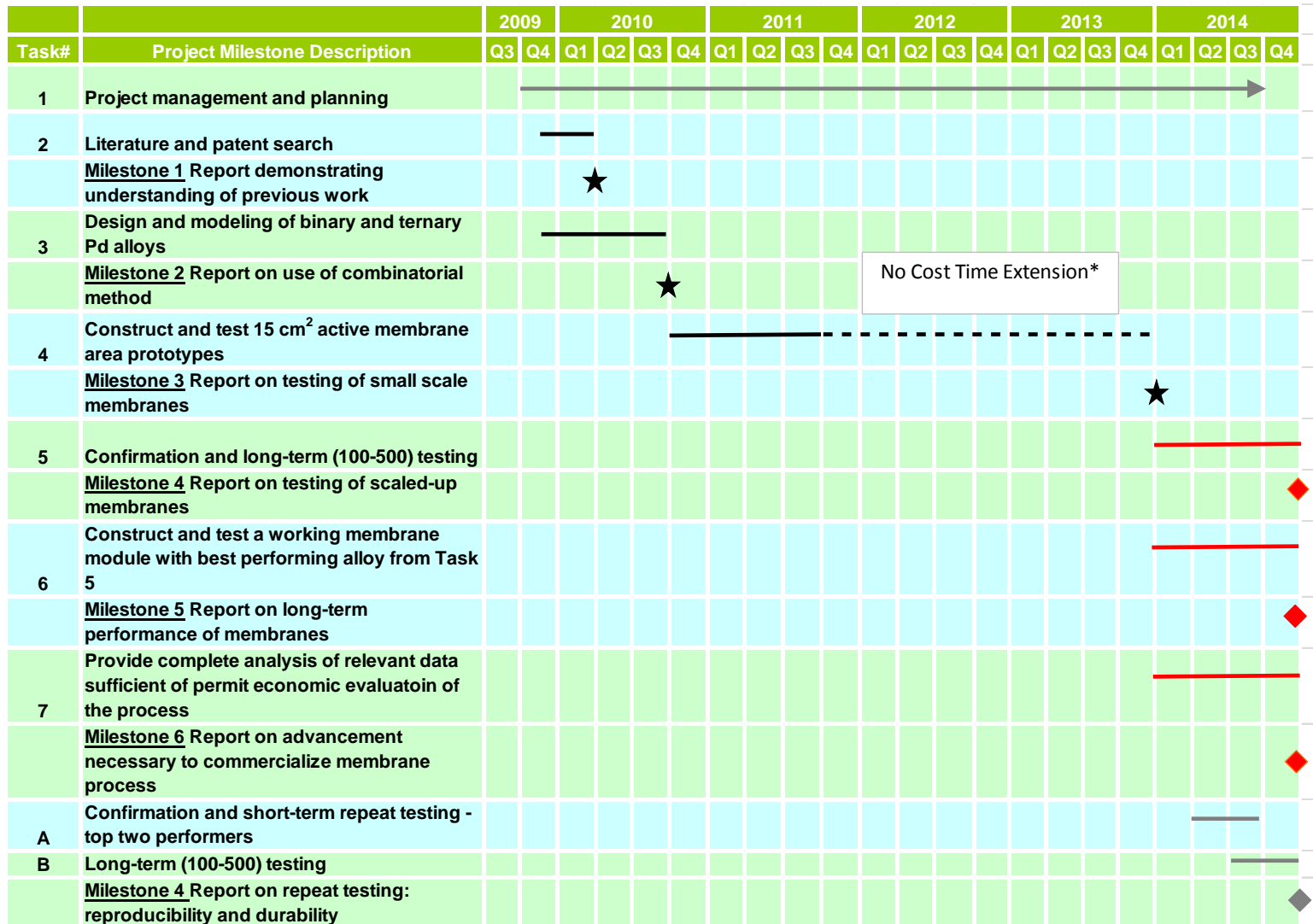
Testing Results at 400°C, $P_{\text{feed}} = 160$ psig



*Note: membranes without the “no carbon” data were tested before the current testing plan was in place.

Project Schedule & Milestones

Completed  Ongoing  #Will not complete 



*Task 4 was delayed, due to 1) negotiation of subcontract with Cornell University, 2) change of subcontractor from Cornell University to Colorado School of Mines to acquire technical capability of alloy membrane deposition on tubular substrate.

#Tasks 5, 6, and 7 were changed to A and B due to time constraints.



Progress and Current Status

Task A

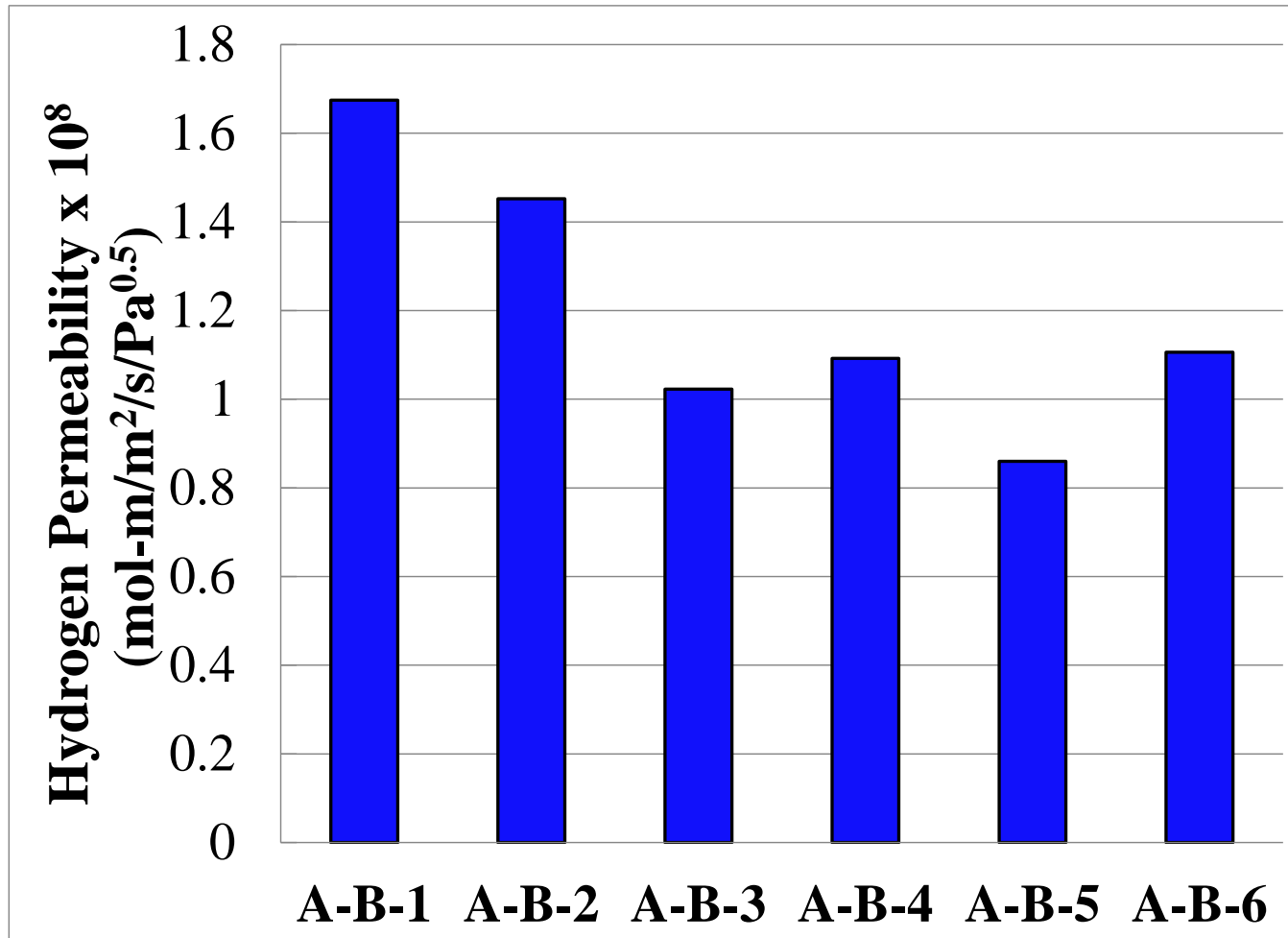
- Fabricate, anneal, and pure-gas test a minimum of three confirmation membranes of top two performing alloys.
- Completion: 100%

Task B

- Repeat mixed-gas testing procedure to confirm initial testing results of at least three membranes.
 - Completion: 67% (two of each membrane have been mixed-gas tested, with one currently being tested)
- Understand long-term (100+ continuous hours in simulated syngas) effects of carbon monoxide and hydrogen sulfide before scaling up.
 - Completion: 0%

Current Work: Repeat Testing of Pd-A-B

Demonstrates the sensitivity of permeability to composition
Permeability data taken at 500°C



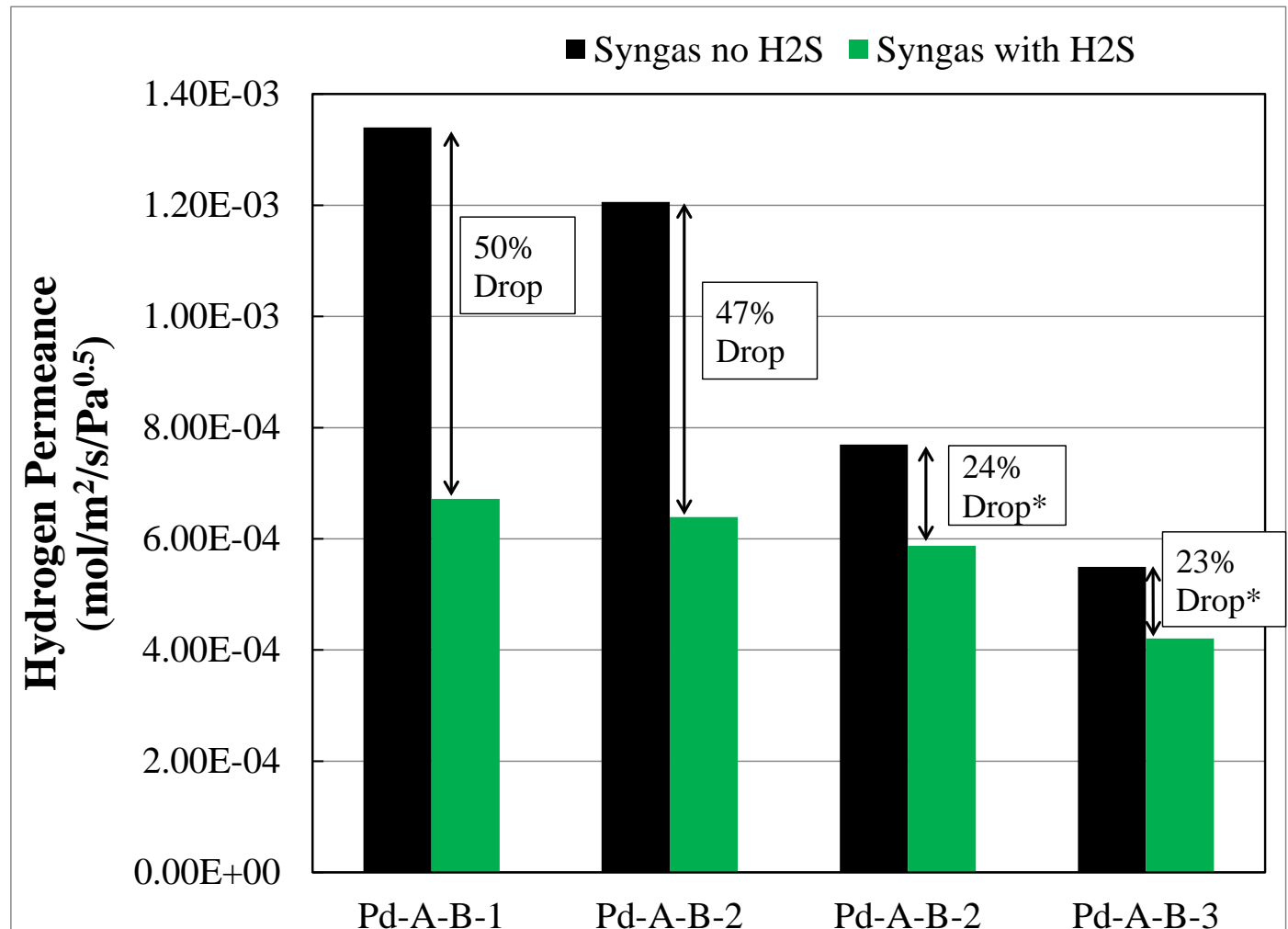
Current Work: Repeat Testing of Pd-A-B

Demonstrates the testing reproducibility (data is taken at 500°C)

Syngas (Test 2B)

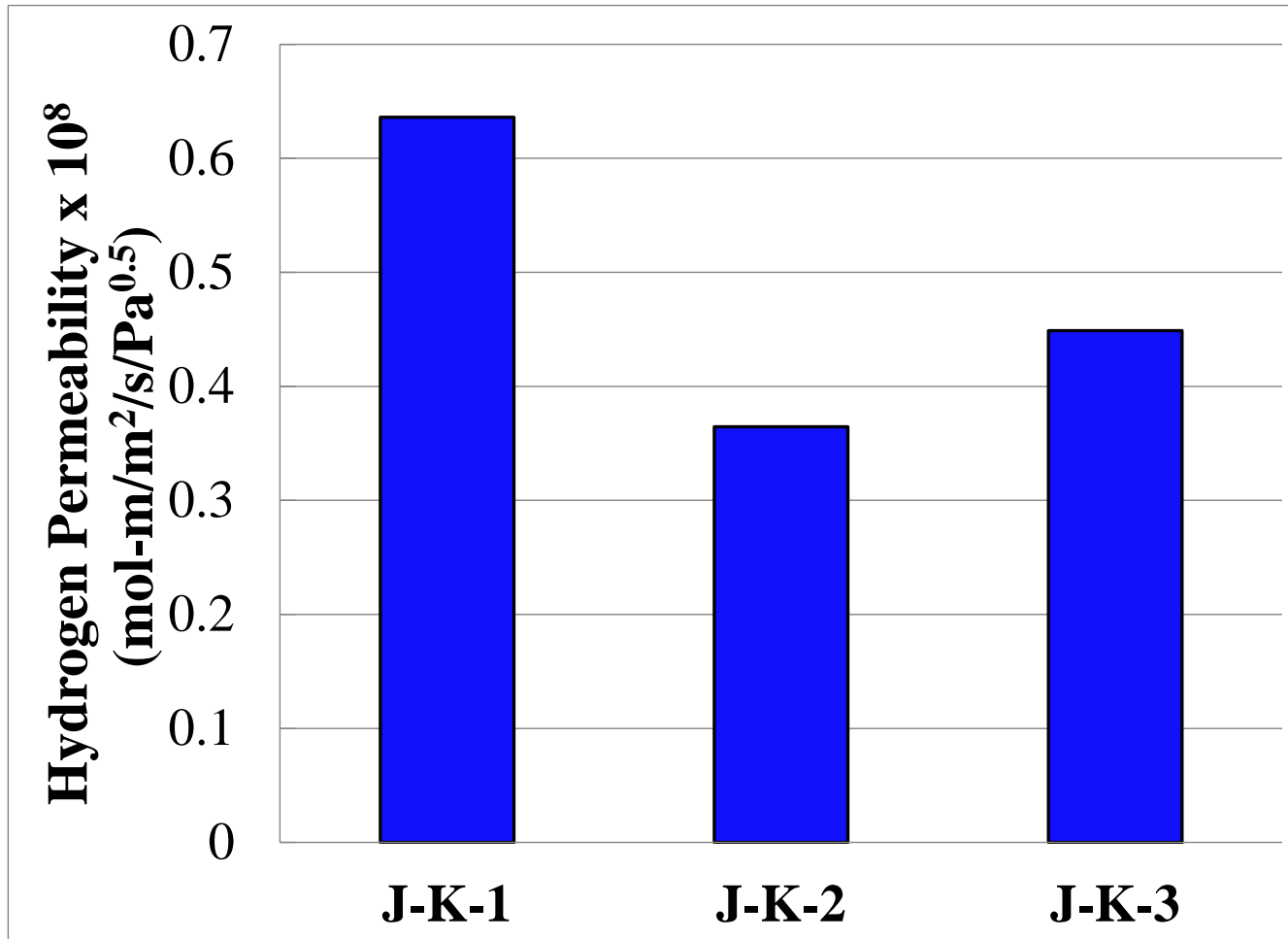
- H₂ 36% → 33%
- CO₂ 11% → 40%
- CO 1.3% = 1.3%
- H₂O 3% → 25%
- N₂ 49% → 0%

*Changed feed gas composition to be DOE test protocol for H₂ separating membranes (test condition 2B, but with 20 ppm H₂S instead of 30)



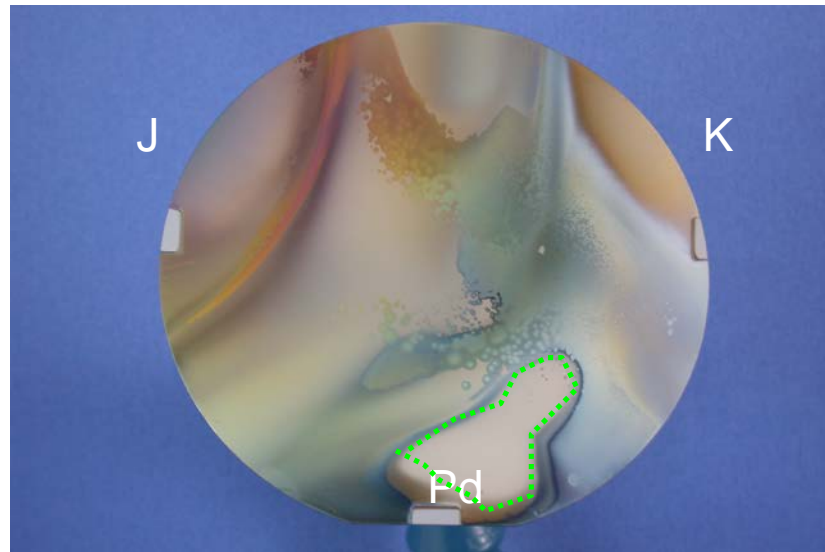
Current Work: Repeat Testing of Pd-J-K

Demonstrates the sensitivity of permeability to composition
Permeability data taken at 500°C



Current Work: Repeat Testing of Pd-J-K

- ❑ Two repeat membranes have been tested in mixed-gas
- ❑ Both failed upon the introduction of H_2S
- ❑ Current hypothesis is that the two repeat membranes are on the edge of what can resist carbon and sulfur species
- ❑ ICP-MS analysis is being used to prove (or disprove) this hypothesis





Plan for Completion

- ❑ Understand fabrication reproducibility
 - ❑ Have completed two repeat tests of each leading candidate with one remaining of Pd-J-K
 - ❑ Use analysis (ICP-MS) to help explain differences

- ❑ Conduct long-term (100+) hour performance test to gain insight to membrane durability of each composition



Acknowledgements

- ❑ U.S. DOE National Energy Technology Laboratory
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- ❑ Pall Corporation
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- ❑ Colorado School of Mines
 - Douglas Way, Colin Wolden, Dan Cooney